Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

3-Ethoxy-4-hydroxybenzaldehyde

Yong Li,[a]* Xinxi Zhang,[a] Jun Zheng[a] and Xiaoling Wang[b]

[a]Department of Environmental Engineering, Anhui University of Technology, Maanshan 243002, People's Republic of China, and [b]Analysis and Testing Central Facility, Anhui University of Technology, Maanshan 243002, People's Republic of China
Correspondence e-mail: farnesene@gmail.com

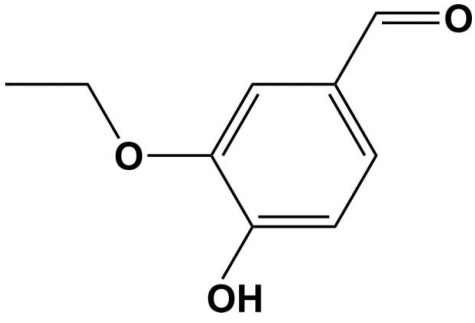
Received 20 September 2008; accepted 22 September 2008

Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.002 Å; R factor = 0.048; wR factor = 0.155; data-to-parameter ratio = 13.2.

The title compound (ethyl vanillin), $C_9H_{10}O_3$, an important food additive and flavouring agent approved by FAO/WHO, has a vanilla odor four times that of vanillin and shows anti-mutagenic activity. There are two molecules in the asymmetric unit, each having a planar conformation and an intramolecular O—H···O bond. Molecules are connected side-by-side, building infinite ribbons along c *via* intermolecular O—H···O hydrogen bonds between the carbonyl and hydroxyl groups. The ribbons are then packed into layers perpendicular to the a axis.

Related literature

For anti-mutagenic activity, see: Ohta *et al.* (1986). For the synthetic method, see: Gradeff & Murayama (1982). For related literature, see: Li (2008).



Experimental

Crystal data

$C_9H_{10}O_3$
M_r = 166.17
Monoclinic, $P2_1/c$
a = 13.7352 (6) Å
b = 14.4140 (6) Å
c = 8.7890 (4) Å
β = 100.742 (3)°
V = 1709.55 (13) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.10 mm^{-1}
T = 296 K
0.50 × 0.50 × 0.40 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: none
16625 measured reflections
3934 independent reflections
2581 reflections with $I > 2\sigma(I)$
R_{int} = 0.025

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.048
$wR(F^2)$ = 0.155
S = 1.01
3934 reflections
297 parameters
All H-atom parameters refined
$\Delta\rho_{max}$ = 0.20 e Å^{-3}
$\Delta\rho_{min}$ = −0.20 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O2—H1···O1^{i}	0.79 (3)	2.03 (3)	2.6951 (17)	142 (3)
O2—H1···O3	0.79 (3)	2.26 (3)	2.6619 (16)	112 (2)
O12—H14···O11^{ii}	0.81 (3)	2.02 (3)	2.7117 (19)	143 (3)
O12—H14···O13	0.81 (3)	2.26 (3)	2.6554 (18)	111 (2)

Symmetry codes: (i) $x, y, z+1$; (ii) $x, y, z-1$.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT-Plus* (Bruker, 2003); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL* software used to prepare material for publication: *SHELXTL*.

The authors thank Professor Tao Zeng for invaluable advice.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DN2380).

References

Bruker (2002). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (2003). *SAINT-Plus*. Bruker AXS Inc., Madison, Wisconsin, USA.
Gradeff, P. S. & Murayama, S. T. (1982). US Patent No. 4 351 962.
Li, Y. (2008). *Chin. J. Struct. Chem.* **27**, 1089-1092.
Ohta, T., Watanabe, M., Watanabe, K., Shirasu, Y. & Kada, T. (1986). *Food Chem. Toxicol.* **24**, 51–54.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

supplementary materials

Acta Cryst. (2008). E**64**, o2008 [**doi:10.1107/S1600536808030419**]

3-Ethoxy-4-hydroxybenzaldehyde

Y. Li, X. Zhang, J. Zheng and X. Wang

Comment

The title compound, an ethyl analogue of vanillin, is an important food additive and flavouring agent approved by FAO/WHO. Its vanilla odor is four times stronger than the flavour of vanillin. Now, it is widely used in food, beverage, cigarette and cosmetics. This synthetic compound was reported to show marked anti-mutagenic activity against mutagenicity induced by 4-nitroquinoline-1-oxide, furylfuramide, captan or methylglyoxal, similar to another report (Li, 2008). It was assumed that the anti-mutagenic activity was due to enhancement of an error-free recombinational repair system (Ohta *et al.*, 1986). But the structure of ethyl vanillin has never been reported. we then report herein its crystal structure determination (Fig.1). The crystal structure consists of layers of planar molecules linked as one-dimensional chains (Fig. 2).

Experimental

One of the synthetic methods was reported by literature(Gradeff & Murayama, 1982). The crude title compound commercially available was recrystallized two times from EtOH/water (1:1) solution, and then colourless block crystals were collected after slow evaporation at room temperature.

Refinement

The structure was solved successfully with direct method. Due to the high quality of diffraction data, R(int) = 0.0252, all H atoms were located in a difference map easily and refined isotropically.

Figures

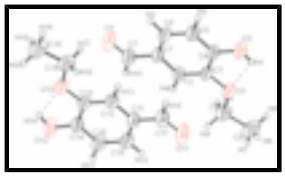


Fig. 1. The molecular structure of the title compound with the atom-labeling scheme. Ellipsoids are drawn at the 50% probability level. H atoms are represented as small spheres of arbitrary radii.

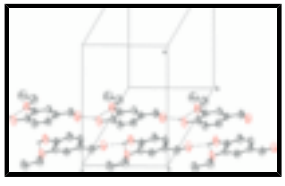


Fig. 2. Partial packing view of the title compound, showing one layer of molecules connected by O—H···O hydrogen bonds (dashed lines). H atoms not involved in hydrogen bondings have been removed for clarity.

3-Ethoxy-4-hydroxybenzaldehyde

Crystal data

$C_9H_{10}O_3$ F_{000} = 704

M_r = 166.17	D_x = 1.291 Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation λ = 0.71073 Å
Hall symbol: -P 2ybc	Cell parameters from 17141 reflections
a = 13.7352 (6) Å	θ = 2.1–27.6º
b = 14.4140 (6) Å	μ = 0.10 mm^{-1}
c = 8.7890 (4) Å	T = 296 K
β = 100.742 (3)º	Block, colourless
V = 1709.55 (13) Å^3	0.50 × 0.50 × 0.40 mm
Z = 8	

Data collection

Bruker SMART CCD area-detector diffractometer	2581 reflections with $I > 2\sigma(I)$
Radiation source: sealed tube	R_{int} = 0.025
Monochromator: graphite	θ_{max} = 27.6º
T = 296 K	θ_{min} = 2.1º
Thin–slice ω scans	h = −17→17
Absorption correction: none	k = −18→18
16625 measured reflections	l = −11→11
3934 independent reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)]$ = 0.048	All H-atom parameters refined
$wR(F^2)$ = 0.155	$w = 1/[\sigma^2(F_o^2) + (0.0737P)^2 + 0.3872P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{max} < 0.001$
3934 reflections	$\Delta\rho_{max}$ = 0.20 e Å^{-3}
297 parameters	$\Delta\rho_{min}$ = −0.20 e Å^{-3}
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Special details

Geometry. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted *R*-factors *wR* and all goodnesses of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating -*R*-factor-obs *etc*. and is not relevant to the choice of reflections for refine-

ment. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.12477 (10)	0.43352 (12)	0.28767 (13)	0.0715 (5)
O2	0.12710 (11)	0.36758 (9)	1.00143 (14)	0.0604 (5)
O3	0.12391 (10)	0.54909 (7)	0.94489 (12)	0.0548 (4)
C1	0.12429 (11)	0.45393 (11)	0.55447 (17)	0.0437 (5)
C2	0.12350 (12)	0.51971 (11)	0.67073 (17)	0.0448 (5)
C3	0.12434 (11)	0.49228 (10)	0.82084 (16)	0.0410 (4)
C4	0.12606 (12)	0.39763 (11)	0.85609 (16)	0.0436 (5)
C5	0.12674 (14)	0.33230 (12)	0.74025 (19)	0.0542 (6)
C6	0.12612 (13)	0.36096 (12)	0.59045 (18)	0.0501 (5)
C7	0.12395 (14)	0.48538 (14)	0.39633 (19)	0.0558 (6)
C8	0.12357 (18)	0.64675 (12)	0.9186 (2)	0.0601 (7)
C9	0.1244 (2)	0.69335 (18)	1.0715 (3)	0.0746 (9)
O11	0.37024 (11)	0.56574 (12)	0.86077 (14)	0.0784 (6)
O12	0.37991 (12)	0.63363 (10)	0.15095 (15)	0.0686 (5)
O13	0.37825 (10)	0.45237 (8)	0.20437 (13)	0.0568 (4)
C11	0.37039 (12)	0.54556 (11)	0.59351 (17)	0.0460 (5)
C12	0.37298 (12)	0.48002 (12)	0.47729 (18)	0.0467 (5)
C13	0.37476 (11)	0.50805 (10)	0.32797 (17)	0.0424 (5)
C14	0.37515 (12)	0.60323 (11)	0.29454 (18)	0.0462 (5)
C15	0.37110 (14)	0.66787 (12)	0.40983 (19)	0.0549 (6)
C16	0.36854 (13)	0.63909 (12)	0.5583 (2)	0.0524 (6)
C17	0.37151 (14)	0.51450 (16)	0.7520 (2)	0.0621 (7)
C18	0.37312 (17)	0.35399 (12)	0.2254 (2)	0.0591 (7)
C19	0.3821 (2)	0.30995 (18)	0.0736 (3)	0.0810 (10)
H1	0.1261 (16)	0.4088 (19)	1.061 (3)	0.080 (7)*
H2	0.1292 (12)	0.2659 (16)	0.769 (2)	0.070 (6)*
H3	0.1271 (13)	0.3143 (15)	0.513 (2)	0.064 (5)*
H4	0.1224 (12)	0.5864 (15)	0.642 (2)	0.065 (6)*
H5	0.0633 (18)	0.6778 (17)	1.111 (3)	0.097 (8)*
H6	0.1828 (19)	0.6754 (18)	1.144 (3)	0.101 (8)*
H7	0.1219 (17)	0.753 (2)	1.052 (3)	0.109 (9)*
H8	0.0602 (16)	0.6625 (15)	0.843 (3)	0.082 (7)*
H9	0.1861 (16)	0.6620 (15)	0.876 (2)	0.079 (6)*
H10	0.1194 (14)	0.5560 (16)	0.378 (2)	0.074 (6)*
H11	0.3722 (13)	0.7337 (16)	0.381 (2)	0.074 (6)*
H12	0.3666 (13)	0.6803 (15)	0.638 (2)	0.064 (5)*
H13	0.3747 (12)	0.4127 (15)	0.507 (2)	0.063 (5)*
H14	0.3786 (17)	0.592 (2)	0.088 (3)	0.095 (8)*
H15	0.3753 (13)	0.4451 (15)	0.767 (2)	0.071 (6)*
H16	0.4291 (15)	0.3369 (14)	0.313 (2)	0.075 (6)*
H17	0.3088 (15)	0.3405 (14)	0.256 (2)	0.071 (6)*
H18	0.383 (2)	0.243 (3)	0.097 (4)	0.138 (11)*

H19	0.325 (2)	0.329 (2)	−0.001 (3)	0.110 (9)*
H20	0.445 (2)	0.3290 (18)	0.042 (3)	0.105 (9)*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0927 (10)	0.0925 (10)	0.0316 (6)	0.0006 (8)	0.0175 (6)	−0.0063 (6)
O2	0.1121 (11)	0.0403 (7)	0.0313 (6)	0.0008 (6)	0.0198 (6)	0.0035 (5)
O3	0.0986 (9)	0.0358 (6)	0.0326 (6)	−0.0012 (6)	0.0188 (5)	−0.0062 (4)
C1	0.0554 (9)	0.0463 (9)	0.0305 (7)	−0.0011 (7)	0.0107 (6)	−0.0016 (6)
C2	0.0664 (10)	0.0357 (8)	0.0333 (8)	−0.0004 (7)	0.0121 (7)	0.0014 (6)
C3	0.0587 (9)	0.0346 (7)	0.0308 (7)	−0.0008 (6)	0.0110 (6)	−0.0041 (6)
C4	0.0660 (10)	0.0364 (8)	0.0297 (7)	0.0002 (7)	0.0125 (6)	0.0016 (6)
C5	0.0917 (13)	0.0337 (8)	0.0389 (8)	0.0001 (8)	0.0169 (8)	−0.0004 (7)
C6	0.0776 (11)	0.0427 (9)	0.0314 (8)	−0.0006 (8)	0.0141 (7)	−0.0081 (7)
C7	0.0722 (11)	0.0635 (12)	0.0327 (8)	−0.0006 (9)	0.0121 (7)	0.0011 (8)
C8	0.0925 (15)	0.0341 (9)	0.0556 (11)	−0.0013 (9)	0.0187 (10)	−0.0071 (8)
C9	0.1020 (19)	0.0521 (12)	0.0708 (15)	0.0004 (12)	0.0193 (13)	−0.0262 (11)
O11	0.1028 (11)	0.0990 (12)	0.0365 (7)	−0.0016 (8)	0.0207 (7)	−0.0071 (7)
O12	0.1270 (12)	0.0454 (7)	0.0374 (7)	0.0021 (7)	0.0254 (7)	0.0043 (6)
O13	0.0963 (9)	0.0380 (6)	0.0384 (6)	0.0026 (6)	0.0188 (6)	−0.0049 (5)
C11	0.0552 (9)	0.0504 (9)	0.0332 (8)	−0.0009 (7)	0.0105 (6)	−0.0028 (7)
C12	0.0628 (10)	0.0391 (9)	0.0393 (8)	−0.0013 (7)	0.0122 (7)	0.0020 (7)
C13	0.0564 (9)	0.0355 (7)	0.0360 (8)	0.0001 (6)	0.0103 (6)	−0.0040 (6)
C14	0.0669 (10)	0.0391 (8)	0.0336 (8)	0.0011 (7)	0.0122 (7)	0.0009 (6)
C15	0.0861 (13)	0.0370 (8)	0.0425 (9)	0.0015 (8)	0.0147 (8)	−0.0047 (7)
C16	0.0704 (11)	0.0473 (9)	0.0408 (9)	0.0010 (8)	0.0134 (8)	−0.0098 (8)
C17	0.0788 (13)	0.0694 (13)	0.0402 (9)	−0.0033 (10)	0.0168 (8)	0.0015 (9)
C18	0.0820 (13)	0.0348 (9)	0.0594 (12)	0.0004 (8)	0.0103 (10)	−0.0076 (8)
C19	0.110 (2)	0.0560 (13)	0.0776 (16)	0.0065 (13)	0.0192 (15)	−0.0263 (12)

Geometric parameters (Å, °)

O1—C7	1.215 (2)	C8—H8	1.02 (2)
O2—C4	1.3464 (19)	C8—H9	1.02 (2)
O3—C3	1.3644 (17)	C9—H6	0.96 (3)
O3—C8	1.426 (2)	C9—H7	0.88 (3)
O2—H1	0.79 (3)	C9—H5	0.99 (3)
O11—C17	1.211 (2)	C11—C12	1.397 (2)
O12—C14	1.349 (2)	C11—C17	1.461 (2)
O13—C13	1.3587 (19)	C11—C16	1.382 (2)
O13—C18	1.434 (2)	C12—C13	1.378 (2)
O12—H14	0.81 (3)	C13—C14	1.403 (2)
C1—C7	1.461 (2)	C14—C15	1.385 (2)
C1—C6	1.376 (2)	C15—C16	1.376 (2)
C1—C2	1.396 (2)	C18—C19	1.503 (3)
C2—C3	1.375 (2)	C12—H13	1.00 (2)
C3—C4	1.398 (2)	C15—H11	0.98 (2)
C4—C5	1.388 (2)	C16—H12	0.923 (19)

C5—C6	1.378 (2)	C17—H15	1.01 (2)
C8—C9	1.501 (3)	C18—H16	1.012 (19)
C2—H4	0.99 (2)	C18—H17	0.99 (2)
C5—H2	0.99 (2)	C19—H18	0.99 (4)
C6—H3	0.96 (2)	C19—H19	0.96 (3)
C7—H10	1.03 (2)	C19—H20	0.99 (3)
O1···O2^{i}	2.6951 (17)	C12···H16	2.71 (2)
O2···C6^{ii}	3.386 (2)	C12···H17	2.822 (19)
O2···O3	2.6619 (16)	C14···H6^{i}	2.92 (3)
O2···O1^{iii}	2.6951 (17)	C16···H16^{ix}	2.82 (2)
O3···O2	2.6619 (16)	C18···H13	2.612 (18)
O11···O12^{iii}	2.7117 (19)	H1···O1^{iii}	2.03 (3)
O12···O11^{i}	2.7117 (19)	H1···O3	2.26 (3)
O12···C16^{iv}	3.372 (2)	H2···H3^{ii}	2.44 (3)
O12···O13	2.6554 (18)	H2···O1^{ii}	2.88 (2)
O13···O12	2.6554 (18)	H3···O2^{v}	2.62 (2)
O1···H3	2.618 (19)	H3···H2^{v}	2.44 (3)
O1···H1^{i}	2.03 (3)	H3···O1	2.618 (19)
O1···H2^{v}	2.88 (2)	H4···H7^{iv}	2.45 (4)
O2···H19^{iii}	2.78 (3)	H4···H8	2.37 (3)
O2···H5^{vi}	2.70 (3)	H4···H9	2.35 (3)
O2···H3^{ii}	2.62 (2)	H4···C8	2.579 (18)
O3···H1	2.26 (3)	H4···H10	2.35 (2)
O11···H9	2.91 (2)	H5···O2^{vi}	2.70 (3)
O11···H14^{iii}	2.02 (3)	H5···C4^{vi}	2.88 (3)
O11···H11^{vii}	2.90 (2)	H6···O12^{iii}	2.76 (3)
O11···H12	2.555 (19)	H6···C14^{iii}	2.92 (3)
O12···H12^{iv}	2.69 (2)	H7···H4^{vii}	2.45 (4)
O12···H6^{i}	2.76 (3)	H8···H4	2.37 (3)
O13···H14	2.26 (3)	H8···C2	2.79 (2)
C1···C11	3.586 (2)	H9···H4	2.35 (3)
C2···C11	3.600 (2)	H9···O11	2.91 (2)
C2···C17	3.349 (3)	H9···C2	2.76 (2)
C2···C7^{viii}	3.340 (3)	H10···H4	2.35 (2)
C3···C17	3.572 (2)	H11···O11^{iv}	2.90 (2)
C3···C7^{viii}	3.599 (2)	H11···H12^{iv}	2.46 (3)
C6···O2^{v}	3.386 (2)	H12···O11	2.555 (19)
C7···C2^{viii}	3.340 (3)	H12···O12^{vii}	2.69 (2)
C7···C12	3.362 (3)	H12···H11^{vii}	2.46 (3)
C7···C3^{viii}	3.599 (2)	H13···C18	2.612 (18)
C11···C2	3.600 (2)	H13···H15	2.33 (2)
C11···C1	3.586 (2)	H13···H16	2.27 (3)
C11···C13^{ix}	3.525 (2)	H13···H17	2.46 (3)

C12···C12^{ix}	3.485 (2)	H13···H18^{ii}	2.38 (5)
C12···C7	3.362 (3)	H14···O11^{i}	2.02 (3)
C12···C13^{ix}	3.571 (2)	H14···O13	2.26 (3)
C13···C12^{ix}	3.571 (2)	H15···H13	2.33 (2)
C13···C11^{ix}	3.525 (2)	H16···C12	2.71 (2)
C16···O12^{vii}	3.372 (2)	H16···H13	2.27 (3)
C17···C3	3.572 (2)	H16···C16^{ix}	2.82 (2)
C17···C2	3.349 (3)	H17···C12	2.822 (19)
C2···H8	2.79 (2)	H17···H13	2.46 (3)
C2···H9	2.76 (2)	H18···H13^{v}	2.38 (5)
C4···H19^{iii}	2.96 (3)	H19···O2^{i}	2.78 (3)
C4···H5^{vi}	2.88 (3)	H19···C4^{i}	2.96 (3)
C8···H4	2.579 (18)		
C3—O3—C8	117.59 (12)	H5—C9—H6	111 (2)
C4—O2—H1	112.7 (19)	C8—C9—H5	110.3 (15)
C13—O13—C18	118.12 (12)	C12—C11—C17	119.54 (16)
C14—O12—H14	113 (2)	C16—C11—C17	120.59 (16)
C6—C1—C7	121.16 (15)	C12—C11—C16	119.86 (14)
C2—C1—C7	119.12 (15)	C11—C12—C13	120.38 (15)
C2—C1—C6	119.72 (14)	O13—C13—C14	114.08 (13)
C1—C2—C3	120.48 (14)	C12—C13—C14	119.18 (14)
C2—C3—C4	119.33 (13)	O13—C13—C12	126.72 (14)
O3—C3—C4	114.26 (12)	O12—C14—C15	118.78 (15)
O3—C3—C2	126.41 (13)	C13—C14—C15	120.14 (14)
O2—C4—C5	118.51 (14)	O12—C14—C13	121.08 (14)
C3—C4—C5	120.10 (13)	C14—C15—C16	120.17 (16)
O2—C4—C3	121.39 (13)	C11—C16—C15	120.24 (16)
C4—C5—C6	119.84 (16)	O11—C17—C11	124.5 (2)
C1—C6—C5	120.53 (15)	O13—C18—C19	106.81 (16)
O1—C7—C1	123.93 (18)	C11—C12—H13	117.8 (10)
O3—C8—C9	107.30 (15)	C13—C12—H13	121.8 (10)
C1—C2—H4	118.2 (10)	C14—C15—H11	117.1 (10)
C3—C2—H4	121.3 (10)	C16—C15—H11	122.7 (10)
C4—C5—H2	118.3 (10)	C11—C16—H12	117.4 (12)
C6—C5—H2	121.9 (10)	C15—C16—H12	122.4 (12)
C1—C6—H3	121.5 (12)	O11—C17—H15	120.6 (10)
C5—C6—H3	118.0 (12)	C11—C17—H15	114.8 (10)
C1—C7—H10	116.4 (10)	O13—C18—H16	106.9 (12)
O1—C7—H10	119.6 (10)	O13—C18—H17	107.3 (12)
O3—C8—H8	107.6 (12)	C19—C18—H16	113.3 (11)
C9—C8—H9	111.4 (11)	C19—C18—H17	112.5 (11)
O3—C8—H9	107.1 (12)	H16—C18—H17	109.6 (15)
C9—C8—H8	110.4 (14)	C18—C19—H18	103 (2)
H8—C8—H9	112.8 (17)	C18—C19—H19	107.0 (17)
C8—C9—H6	109.8 (16)	C18—C19—H20	110.5 (15)
C8—C9—H7	105.7 (17)	H18—C19—H19	113 (2)
H5—C9—H7	107 (2)	H18—C19—H20	111 (2)

H6—C9—H7	113 (2)	H19—C19—H20	112 (2)
C8—O3—C3—C4	179.15 (16)	C3—C4—C5—C6	0.3 (3)
C8—O3—C3—C2	−0.7 (3)	O2—C4—C5—C6	−179.73 (17)
C3—O3—C8—C9	−179.32 (17)	C4—C5—C6—C1	−0.4 (3)
C13—O13—C18—C19	−178.06 (17)	C16—C11—C12—C13	0.7 (3)
C18—O13—C13—C14	−176.92 (16)	C12—C11—C17—O11	179.41 (18)
C18—O13—C13—C12	4.4 (3)	C16—C11—C17—O11	0.5 (3)
C6—C1—C2—C3	−0.1 (2)	C17—C11—C16—C15	177.65 (17)
C7—C1—C6—C5	179.79 (17)	C12—C11—C16—C15	−1.2 (3)
C2—C1—C6—C5	0.3 (3)	C17—C11—C12—C13	−178.19 (16)
C2—C1—C7—O1	179.85 (18)	C11—C12—C13—O13	179.34 (16)
C7—C1—C2—C3	−179.64 (16)	C11—C12—C13—C14	0.8 (2)
C6—C1—C7—O1	0.3 (3)	O13—C13—C14—C15	179.53 (16)
C1—C2—C3—C4	0.0 (2)	C12—C13—C14—O12	177.97 (16)
C1—C2—C3—O3	179.87 (16)	C12—C13—C14—C15	−1.7 (2)
O3—C3—C4—O2	0.1 (2)	O13—C13—C14—O12	−0.8 (2)
C2—C3—C4—O2	179.89 (15)	C13—C14—C15—C16	1.2 (3)
O3—C3—C4—C5	−179.95 (16)	O12—C14—C15—C16	−178.48 (17)
C2—C3—C4—C5	−0.1 (2)	C14—C15—C16—C11	0.3 (3)

Symmetry codes: (i) *x*, *y*, *z*−1; (ii) *x*, −*y*+1/2, *z*+1/2; (iii) *x*, *y*, *z*+1; (iv) *x*, −*y*+3/2, *z*−1/2; (v) *x*, −*y*+1/2, *z*−1/2; (vi) −*x*, −*y*+1, −*z*+2; (vii) *x*, −*y*+3/2, *z*+1/2; (viii) −*x*, −*y*+1, −*z*+1; (ix) −*x*+1, −*y*+1, −*z*+1.

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O2—H1···O1^{iii}	0.79 (3)	2.03 (3)	2.6951 (17)	142 (3)
O2—H1···O3	0.79 (3)	2.26 (3)	2.6619 (16)	112 (2)
O12—H14···O11^{i}	0.81 (3)	2.02 (3)	2.7117 (19)	143 (3)
O12—H14···O13	0.81 (3)	2.26 (3)	2.6554 (18)	111 (2)

Symmetry codes: (iii) *x*, *y*, *z*+1; (i) *x*, *y*, *z*−1.

Fig. 1

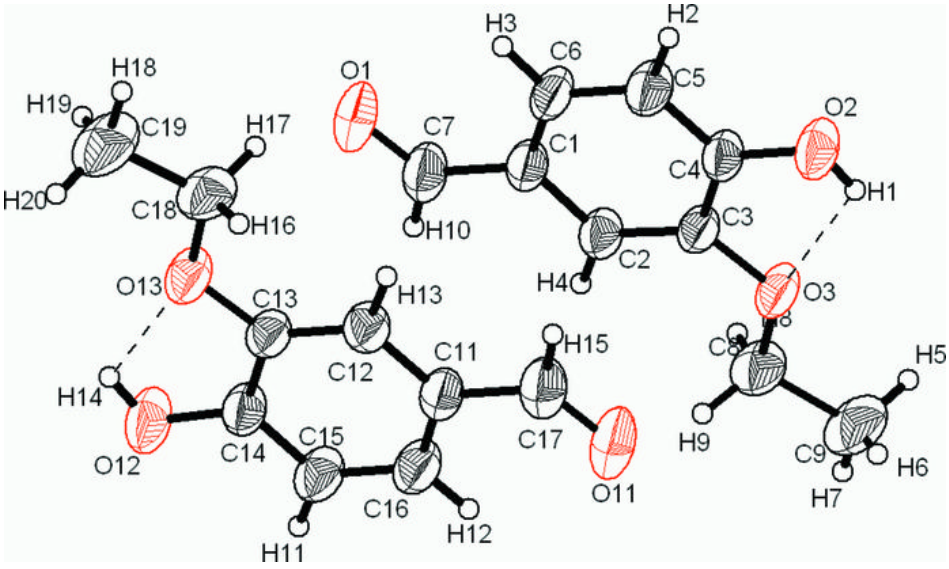

H3
H2
C6
C5
H18
O1
O2
H19
C19
H17
C7
C1
C4
H1
H20
C18
H16
H10
C3
C2
H4
O13
C13
H13
O3
C12
C11
H15
H5
H14
C17
H9
C9
H6
O12
C14
C15
O11
C16
H7
H11
H12

Fig. 2